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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.004 Å R factor = 0.044 wR factor = 0.146 Data-to-parameter ratio = 12.2

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3-Phenyl-4a,5-dihydro-1,2,4-triazolo[3,4-*b*]-[1,3]benzothiazine

In the crystal structure of the title compound, $C_{15}H_{11}N_3S$, the thiazine ring adopts a boat conformation. The dihedral angle between the triazole and phenyl rings is 34.3 (1)°. The packing of the molecule is stabilized by π - π interactions.

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Comment

Triazole derivatives exhibit antimicrobial (Habib *et al.*, 1997), antiviral (Ergen *et al.*, 1996), anti-HIV-1 (Invidiata *et al.*, 1996), antifungal, antimycobacterial and anticonvulsant (Gulerman *et al.*, 1997) activities. Triazole is also a highly potent eosinophilia inhibitor (Naito *et al.*, 1996), fungicide (Croften, 1996) and herbicide (Tada *et al.*, 1995). Some of the benzothiazine derivatives are most active against the Grampositive *Bacillus subtilis* (Armenise *et al.*, 1991). Triazole derivatives also possesses neuroleptic activities (Grol & Rollema, 1975). The X-ray crystal structure analysis of the title compound, (I), was carried out as part of our studies on triazole derivatives.



The N-N, C-N and C-S bond distances in the fused ring system are comparable with those found in a related structure, *viz.* 3-amino-6,7-dihydro-6-hydroxy-5*H*-1,2,4-triazolo[3,4-*b*]-[1,3]thiazine (Glowka, 1994). The S1-C1 [1.736 (3) Å], C1-N1 [1.360 (4) Å] and C1-N3 [1.314 (4) Å] bond distances clearly reflect the resonance of the thiourea system (Valle *et al.*, 1970). The C1-S1-C2 angle of 97.5 (1)° suggests that the S atom uses only *p*-orbitals to form bonds with atom C1 and C2. In order to minimize the steric repulsion between the H atoms at C8 and C11, the relevant bond angles, *viz.* C8-N1-C9 and N1-C9-C10, are widened.

In the title compound, (I) (Fig. 1), the benzene ring (A) is planar, with a maximum deviation of 0.010 (4) Å for C5. The thiazine ring (B) is in a boat conformation, with puckering parameters (Cremer & Pople, 1975) $q_2 = 0.547$ (2), $Q_3 =$ -0.054 (3), $Q_T = 0.549$ (2), $\Phi = 124^{\circ}$ and $\theta = 96^{\circ}$, and asymmetry parameters (Nardelli, 1983) $\Delta_s(C8) = 0.024$ (1) and $\Delta_s(N1-C1) = 0.030$ (1). Atoms S1 and C8 deviate by 0.552 (1) and 0.400 (3) Å, respectively, from the weighted least-squares plane through the remaining four atoms C1, N1, C7 and C2. The dihedral angle between the benzene (A), triazole (C) and phenyl (D) rings are: A/C 38.5 (1) A/D 70.9 (1) and C/D 34.3 (1)°.



Figure 1

The molecular structure, with the atomic numbering scheme. Probability displacement ellipsoids are drawn at the 50% level.

In addition to van der Waals interactions, $\pi - \pi$ interactions are possibly involved to ensure cohesion between the molecules. There is a π - π interaction between the triazole ring at (x, y, z) and the phenyl ring at $(\frac{1}{2} - x, -\frac{1}{2} + y, z)$, the centroids of the two rings being separated by 3.531 (2) Å.

Experimental

The title compound was synthesized by irradiation (254 nm) of 4-(2chlorobenzyl)-5-phenyl-1,2,4-triazole-3-thione, according to a literature method (Park et al., 1999), using CH₃CN/NaOH as solvent. After completion of the reaction, removal of solvent and chromatographic separation on a silica-gel column by elution with a petroleum etherethyl acetate (1:1) mixture afforded the title compound.

Crystal data

C15H11N3S Mo $K\alpha$ radiation $M_r = 265.33$ Orthorhombic, Pbca reflections a = 11.5674 (15) Å $\theta = 5 - 20^{\circ}$ $\mu = 0.24 \text{ mm}^{-1}$ b = 10.1948 (19) Åc = 21.433 (3) Å T = 293 (2) KV = 2527.5 (7) Å³ Block, colorless Z = 8 $D_x = 1.395 \text{ Mg m}^{-3}$ Data collection $\theta_{\rm max} = 25.0^{\circ}$ $h = -2 \rightarrow 13$ Enraf-Nonius CAD-4 diffractometer $k = -2 \rightarrow 12$ $\omega/2\theta$ scans Absorption correction: none $l = -25 \rightarrow 5$ 2197 measured reflections 2110 independent reflections 1457 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.015$ Refinement Refinement on F^2

 $R[F^2 > 2\sigma(F^2)] = 0.044$ $wR(F^2) = 0.146$ S = 0.882110 reflections 173 parameters H-atom parameters constrained

Cell parameters from 26 $0.46 \times 0.26 \times 0.23$ mm

2 standard reflections every 200 reflections frequency: 120 min intensity decay: <1.5%

 $w = 1/[\sigma^2(F_o^2) + (0.1P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.19 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\rm min} = -0.19 \ {\rm e} \ {\rm \AA}^{-3}$ Extinction correction: SHELXL97 Extinction coefficient: 0.0065 (13)



Figure 2

A packing diagram of the molecule, viewed down the *a* axis. $\pi - \pi$ interactions are indicated by dashed lines.

Table 1

Selected geometric parameters (Å, °).

S1-C1	1.736 (3)	N1-C1	1.360 (4)
S1-C2	1.773 (3)	C9-C10	1.473 (4)
N1-C9	1.364 (3)	N2-N3	1.393 (4)
C1-S1-C2	97.49 (14)	N2-C9-N1	110.0 (3)
C9-N1-C8	131.6 (2)	N1-C9-C10	125.6 (2)
C1-N1-C8	123.6 (2)	C9-N2-N3	108.1 (2)

The H atoms were positioned geometrically and were treated as riding on their parent C atoms, with aromatic C-H distances of 0.93 Å and other C-H distances of 0.97 Å,

Data collection: CAD-4 EXPRESS (Enraf-Nonius, 1994); cell refinement: CAD-4 EXPRESS; data reduction: XCAD4 (Harms & Wocadlo, 1995); program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ZORTEP (Zsolnai, 1997) and PLATON (Spek, 1990); software used to prepare material for publication: SHELXL97 and PARST (Nardelli, 1995).

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